



# CAREER POINT

## KVPY PAPER-2020 (STREAM SA)

### Part - I

### One - Mark Questions

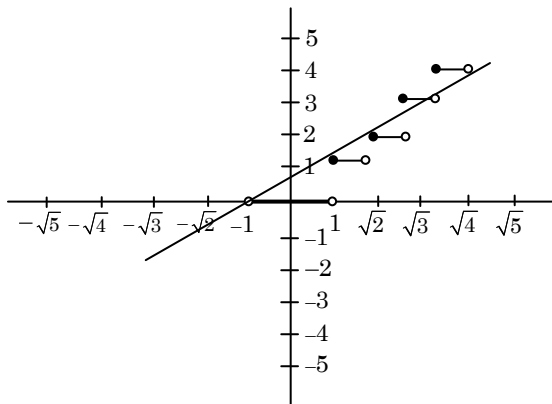
Date : 31 / 01 / 2021

## MATHEMATICS

1. Let  $[x]$  be the greatest integer less than or equal to  $x$ , for a real number  $x$ . Then the equation  $[x^2] = x + 1$  has  
 (A) two solutions                      (B) one solution                      (C) no solution                      (D) more than two solutions

Ans. [C]

Sol.



From the graph it is clear that the equation has no-solution.

2. Let  $p_1(x) = x^3 - 2020x^2 + b_1x + c_1$  and  $p_2(x) = x^3 - 2021x^2 + b_2x + c_2$  be polynomials having two common roots  $\alpha$  and  $\beta$ . Suppose there exist polynomials  $q_1(x)$  and  $q_2(x)$  such that  $p_1(x)q_1(x) + p_2(x)q_2(x) = x^2 - 3x + 2$ . Then the correct identity is  
 (A)  $p_1(3) + p_2(1) + 4028 = 0$                       (B)  $p_1(3) + p_2(1) + 4026 = 0$   
 (C)  $p_1(2) + p_2(1) + 4028 = 0$                       (D)  $p_1(1) + p_2(2) + 4028 = 0$

Ans. [A]

Sol.

$$p_1(x)q_1(x) + p_2(x)q_2(x) = x^2 - 3x + 2$$

$$p_1(x) - p_2(x) = x^2 + (b_1 - b_2)x + (c_1 - c_2)$$

$$\Rightarrow q_1(x) = 1 \text{ \& } q_2(x) = -1$$

$$p_1(x) = x^3 - 2020x^2 + b_1x + c_1 \begin{cases} 1 \\ 2 \\ t \end{cases}$$

$$t + 3 = 2020 \Rightarrow t = 2017$$

$$p_1(x) = (x - 1)(x - 2)(x - 2017)$$

$$\text{Similarly } p_2(x) = (x - 1)(x - 2)(x - 2018)$$

$$(A) \quad p_1(3) + p_2(1) + 4028 = 0$$

$$p_1(3) = -4028$$

$$p_2(1) = 0$$

Hence it is true

3. Suppose  $p, q, r$  are positive rational numbers such that  $\sqrt{p} + \sqrt{q} + \sqrt{r}$  is also rational. Then

- (A)  $\sqrt{p}, \sqrt{q}, \sqrt{r}$  are irrational  
 (B)  $\sqrt{pq}, \sqrt{pr}, \sqrt{qr}$  are rational, but  $\sqrt{p}, \sqrt{q}, \sqrt{r}$  are irrational  
 (C)  $\sqrt{p}, \sqrt{q}, \sqrt{r}$  are rational  
 (D)  $\sqrt{pq}, \sqrt{pr}, \sqrt{qr}$  are irrational

Ans. [C]

Sol.  $\sqrt{p} + \sqrt{q} + \sqrt{r} \in \mathbb{Q}, p, q, r \in \mathbb{Q}$

let  $\sqrt{p} + \sqrt{q} + \sqrt{r} = t$

$\sqrt{p} + \sqrt{q} = t - \sqrt{r}$

$p + q + 2\sqrt{pq} = t^2 + r - 2t\sqrt{r}$

$\sqrt{pq} + t\sqrt{r} \in \mathbb{Q} = \lambda, \lambda \in \mathbb{Q}$

$\sqrt{pq} = \lambda - t\sqrt{r}$

$pq = \lambda^2 + t^2r - 2\lambda t\sqrt{r}$

$\Rightarrow \sqrt{r} \in \mathbb{Q}$  similarly  $\sqrt{p}$  and  $\sqrt{q} \in \mathbb{Q}$

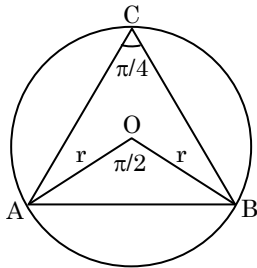
hence  $\sqrt{p}, \sqrt{q}, \sqrt{r} \in \mathbb{Q}$

4. Let  $A, B, C$  be three points on a circle of radius 1 such that  $\angle ACB = \frac{\pi}{4}$ . Then the length of the side  $AB$  is-

- (A)  $\sqrt{3}$  (B)  $\frac{4}{3}$  (C)  $\frac{3}{\sqrt{2}}$  (D)  $\sqrt{2}$

Ans. [D]

Sol.



Let  $O$  be the centre of the circle

In  $\triangle OAB$

$AB = \sqrt{2}r$  and  $r = 1$

$AB = \sqrt{2}r$  and  $r = 1$

$\Rightarrow AB = \sqrt{2}$

5. Let  $x$  and  $y$  be two positive real numbers such that  $x + y = 1$ . Then the minimum value of  $\frac{1}{x} + \frac{1}{y}$  is-

- (A) 2 (B)  $\frac{5}{2}$  (C) 3 (D) 4

Ans. [D]

**Sol.**  $x + y = 1$  and  $x, y > 0$

Apply  $AM \geq HM$

$$\frac{x+y}{2} \geq \frac{2}{\frac{1}{x} + \frac{1}{y}}$$

$$\Rightarrow \frac{1}{x} + \frac{1}{y} \geq 4$$

6. Let ABCD be a quadrilateral such that there exists a point E inside the quadrilateral satisfying  $AE = BE = CE = DE$ . Suppose  $\angle DAB, \angle ABC, \angle BCD$  is an arithmetic progression. Then the median of the set  $\{\angle DAB, \angle ABC, \angle BCD\}$  is-

(A)  $\frac{\pi}{6}$

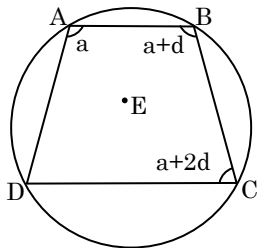
(B)  $\frac{\pi}{4}$

(C)  $\frac{\pi}{3}$

(D)  $\frac{\pi}{2}$

**Ans.** [D]

**Sol.**



$$AE = BE = CE = DE$$

$$\angle DAB, \angle ABC, \angle BCD \rightarrow AP$$

$$\text{Let } \angle DAB = a$$

$$\angle ABC = a + d$$

$$\angle BCD = a + 2d$$

Since  $AE = BE = CE = DE$  so ABCD is cyclic quadrilateral

$$\text{Hence } \angle DAB + \angle DCB = 180^\circ$$

$$2a + 2d = 180^\circ \Rightarrow a + d = 90^\circ$$

so median of  $\{a, a + d, a + 2d\}$  is  $a + d = 90^\circ$

7. The number of ordered pairs  $(x, y)$  of positive integers satisfying  $2^x + 3^y = 5^{xy}$  is

(A) 1

(B) 2

(C) 5

(D) infinite

**Ans.** [A]

**Sol.**  $2^x + 3^y = 5^{xy}$

clearly  $x = y = 1$  satisfy the relation

Take  $x > y$

$$3^x > 3^y$$

$$2^x + 3^x > 2^x + 3^y$$

$$2^x + 3^x > 5^{xy}$$

this is false

$$5^{xy} > 5^x = (2 + 3)^x > 2^x + 3^x$$

Similarly there is no solution for  $x < y$

Hence  $x = y$ , hence only  $x = y = 1$  satisfy the given equation ( $2^x + 3^x = 5^{x^2}$  is not true for  $x \in \mathbb{N} - \{1\}$ )

8. In the integers from 1 to 2021 are written as a single integer like 123....91011.....20202021, then the 2021<sup>st</sup> digit (counted from the left) in the resulting number is-

(A) 0

(B) 1

(C) 6

(D) 9

**Ans.** [B]

Sol. 1 2 3 4 5 ... 9 10 11 12 13 ... 99 100 101

find the 2021 term

double digit =  $90 \times 2$

1 2 3 4 5 6 7 8 9 10 11 12 .....99 100 101  
 9-digit 180-digit

we need 2021<sup>st</sup> digit

Till two digit number we have 189 digit

we need  $2021 - 189 = 1832$  digit

triple digit =  $\frac{1832}{3} = 610 \times 3 + 2$

we take 610 three digit number

100, 101, ....., 709

1 2 3 ..... 9 10 11 12 13 .... 99 100 101 .... 709 71 → 2021<sup>st</sup> digit  
 9 digit + 180 digit + 1830 digit 2021<sup>st</sup> digit  
 Total =  $9 + 180 + 1830 = 2019$

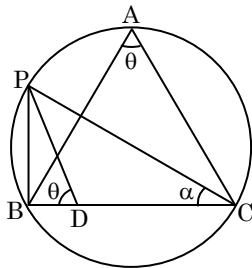
Ans = 1

9. In a triangle ABC, a point D is chosen on BC such that  $BD : DC = 2 : 5$ . Let P be a point on the circumcircle ABC such that  $\angle PDB = \angle BAC$ . Then  $PD : PC$  is

- (A)  $\sqrt{2} : \sqrt{5}$  (B)  $2 : 5$  (C)  $2 : 7$  (D)  $\sqrt{2} : \sqrt{7}$

Ans. [D]

Sol.



$$\frac{BD}{DC} = \frac{2}{5}$$

$$\angle PDB = \angle BAC = \theta$$

$$\text{let } \angle PCD = \alpha$$

$$\Rightarrow \angle DPC = \theta - \alpha$$

$$\angle BAC = \angle BPC = \theta$$

(angle in the same segment)

$$\Rightarrow \angle BPD = \theta - (\theta - \alpha) = \alpha$$

so  $\triangle PCB \sim \triangle PDB$

$$\frac{PC}{DP} = \frac{BC}{PB} = \frac{PB}{BD}$$

$$\left(\frac{PC}{DP}\right)^2 = \frac{BC}{PB} \times \frac{PB}{BD} = \frac{BC}{BD}$$

$$\frac{PC}{DP} = \sqrt{\frac{BC}{BD}} = \sqrt{\frac{7\lambda}{2\lambda}} = \frac{\sqrt{7}}{\sqrt{2}}$$

$$\frac{DP}{PC} = \frac{\sqrt{2}}{\sqrt{7}}$$

10. Let  $[x]$  be the greatest less than or equal to  $x$ , for a real number  $x$ . Then the following sum

$$\left[ \frac{2^{2020} + 1}{2^{2018} + 1} \right] + \left[ \frac{3^{2020} + 1}{3^{2018} + 1} \right] + \left[ \frac{4^{2020} + 1}{4^{2018} + 1} \right] + \left[ \frac{5^{2020} + 1}{5^{2018} + 1} \right] + \left[ \frac{6^{2020} + 1}{6^{2018} + 1} \right]$$

- (A) 80 (B) 85 (C) 90 (D) 95

Ans. [B]

Sol.  $\frac{2^{2020} + 1}{2^{2018} + 1} = \frac{4 \cdot 2^{2018} + 1}{2^{2018} + 1} = \frac{4(2^{2018} + 1) - 3}{2^{2018} + 1}$

$$4 - \frac{3}{2^{2018} + 1} = t, 3 < t < 4$$

Now  $\left[ \frac{2^{2020} + 1}{2^{2018} + 1} \right] = 3$

similarly  $\frac{3^{2020} + 1}{3^{2018} + 1} = \frac{9 \cdot 3^{2018} + 1}{3^{2018} + 1} = 9 - \frac{8}{3^{2018} + 1}$

$$\left[ \frac{3^{2020} + 1}{3^{2018} + 1} \right] = 8$$

similarly  $\left[ \frac{n^{2020} + 1}{2^{2018} + 1} \right] = n^2 - 1$

$$(2^2 - 1) + (3^2 - 1) + (4^2 - 1) + (5^2 - 1) + (6^2 - 1) = 3 + 8 + 15 + 24 + 35 = 85$$

11. Let  $r$  be the remainder when  $2021^{2020}$  is divided by  $2020^2$ . Then  $r$  lies between

- (A) 0 and 5 (B) 10 and 15 (C) 20 and 100 (D) 107 and 120

Ans. [A]

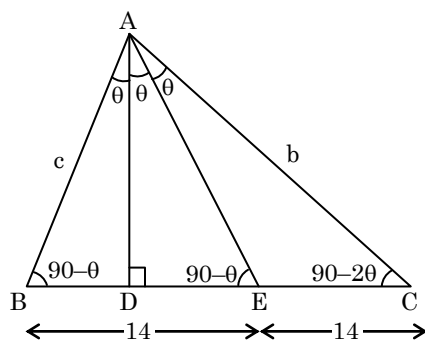
Sol.  $(2021)^{2020} = (1 + 2020)^{2020}$   
 ${}^{2020}C_0 + {}^{2020}C_1 \cdot 2020 + {}^{2020}C_2 \cdot 2020^2 + \dots + {}^{2020}C_{2020} \cdot 2020^{2020}$   
 $1 + (2020)^2 + {}^{2020}C_2 \cdot 2020^2 + \dots + {}^{2020}C_{2020} \cdot 2020^{2020}$   
 $1 + (2020)^2 (1 + {}^{2020}C_2 + \dots + (2020)^{2018})$   
 $1 + (2020)^2 \cdot \lambda$   
 Hence  $(2021)^{2020} = \lambda(2020)^2 + 1$   
 Hence remainder = 1

12. In a triangle ABC, the altitude AD and the median AE divide  $\angle A$  into three equal parts. If  $BC = 28$ , then the nearest integer to  $AB + AC$  is-

- (A) 38 (B) 37 (C) 36 (D) 33

Ans. [A]

Sol.



$\triangle ABE$  is isosceles  $\Rightarrow BD = DE = 7$

$$\triangle ADC : \tan(90 - 2\theta) = \frac{AD}{21} \quad \dots(1)$$

$$\triangle ADE : \tan(90 - \theta) = \frac{AB}{7} \quad \dots(2)$$

$$\text{Divide } \frac{\tan \theta}{\tan 2\theta} = \frac{1}{3} \Rightarrow \frac{1 - \tan^2 \theta}{2} = \frac{1}{3}$$

$$1 - \tan 2\theta = \frac{2}{3} \Rightarrow \tan \theta = \frac{1}{\sqrt{3}} \Rightarrow \theta = 30^\circ$$

$$\triangle ABD : \cos(90 - \theta) = \frac{BD}{C} = \sin \theta$$

$$C = 7 \operatorname{cosec} \theta = 14$$

$$\triangle ADC : \cos(90 - 2\theta) = \frac{DC}{b} = \sin 2\theta$$

$$b = 21 \operatorname{cosec} 2\theta = 21 \operatorname{cosec} \frac{\pi}{3}$$

$$b = \frac{42}{\sqrt{3}} = 14\sqrt{3}$$

$$b + c = 14\sqrt{3} + 14$$

$$[b + c] = 38$$

13. The number of permutations of the letters  $a_1, a_2, a_3, a_4, a_5$  in which the first letter  $a_1$  does not occupy the first position (from the left) and the second letter  $a_2$  does not occupy the second position (from the left) is-

(A) 96                                      (B) 78                                      (C) 60                                      (D) 42

Ans. [B]

Sol.

(When  $a_1$  does not occupy its position)                      (When  $a_1$  does not occupy its position but  $a_2$  occupy its second position)

$$\begin{array}{ccccccc} 4 & \times & 4! & - & 3 & \times & 3! & = & 78 \\ \text{a}_1 \text{ can occupy} & & \text{Remaining} & & \text{a}_1 \text{ can occupy} & & \text{Remaining} & & \\ \text{any position} & & \text{4 letter can} & & \text{3-position} & & \text{three person} & & \\ \text{except 1}^{\text{st}} & & \text{be arranged} & & \text{except 1}^{\text{st}} \text{ and 2}^{\text{nd}} & & \text{arranged in} & & \\ & & \text{in 4-position} & & & & \text{3-position} & & \end{array}$$

14. There are  $m$  books in black cover and  $n$  books in blue cover, and all books are different. The number of ways these  $(m + n)$  books can be arranged on a shelf so that all the book in black cover are put side by side is

(A)  $m! n!$                                       (B)  $m!(n + 1)!$                                       (C)  $(n + 1)!$                                       (D)  $(m + n)!$

Ans. [B]

Sol. Put all block cover books together

$$A_1 A_2 A_3 \dots A_m \rightarrow \alpha$$

$$\text{Total number of books} = n + 1$$

These books be arranged in  $(n + 1)!$  ways and  $m$  books be arranged  $m!$  ways

$$\text{No. of way} = m! (n + 1)!$$

15. A 5 digit number  $abcde$ , when multiplied by 9, gives the 5-digit number  $edcba$ . The sum of the digits in the number is-
- (A) 18 (B) 27 (C) 36 (D) 45

Ans. [B]

Sol.  $abcde \times 9 = edcba$

surely  $a = 1$

$$\Rightarrow 1bcde \times 9 = edcb1$$

9e last digit is 1  $\Rightarrow e = 9$

$$\Rightarrow 1bcd9 \times 9 = 9dcb1$$

9 multiply by  $b \Rightarrow b$  has to  $\{0, 1\}$  otherwise RHS is a six digit number

C-1 Take  $b = 0$

$$10cd9 \times 9 = 9dc01$$

$9d + 8 = 0 \rightarrow$  (last digit has to be zero)

$$\Rightarrow d = 8$$

$$10c89 \times 9 = 98c01$$

Now  $98c01$  is divisible by 9  $\Rightarrow$  sum of digit divisible by 9  $\Rightarrow c = 0, 9$

take  $c = 0, 10089 \times 9 = 90801$  (rejected)

take  $c = 9, 1089 \times 9 = 9801$

$a = 1, b = 0, c = 9, d = 8, e = 9$

Sum = 27

C-2 take  $b = 1$

$$11cd9 \times 9 = 9dc11$$

$9d + 8 = 1 \Rightarrow d = 7$

$$9 \times 11c79 = 97c11$$

This cannot be true for  $c \in \{0, 1, 2, \dots, 9\}$

**Alternate Solution**

$$9(a \times 10^4 + b \times 10^3 + c \times 10^2 + b \times 10 + e)$$

$$= e \times 10^4 + d \times 10^3 + c \times 10^2 + b \times 10 + a$$

$$89999a + 8990b + 800c - 910d - 9991e = 0$$

for max. value of 'a'

put  $b = c = 0$  and  $d = e = 9$

$$a = \frac{98109}{89999} \Rightarrow a \text{ will be } 1$$

$$\therefore 89999 + 8990b + 800c - 910d - 9991e = 0$$

for max. value of b

$$\therefore b = \frac{8110}{8990} \Rightarrow b \text{ will be } 0$$

$$\therefore 89999 + 800c - 910d - 9991e = 0$$

for max. value of c

put  $d = e = 9 \Rightarrow c > 10$  (not possible)

put  $d = e = 8$  (not possible)

put  $d = 9, e = 9$  (not possible)

put  $d = 8, e = 9$

$$\Rightarrow c = \frac{7200}{800} \Rightarrow c = 9$$

$\therefore$  number is 10989

## PHYSICS

16. A mouse jumps off from the 15<sup>th</sup> floor of a high-rise building and lands 12 m from the building. Assume that each floor is of 3m height. The horizontal speed with which the mouse jumps is closest to-

(A) 0 (B) 5 kmph (C) 10 kmph (D) 15 kmph

Ans. [D]

Sol. Time of fall =  $\sqrt{\frac{2h}{g}} = \sqrt{\frac{2 \times 45}{10}}$

$$t = 3 \text{ sec}$$

horizontal distance = horizontal velocity  $\times$  time

$$12 = v \times 3$$

$$v = 4 \text{ m/s}$$

$$= 4 \times \frac{18}{5} \text{ km/hr}$$

$$v = 14.4 \text{ km/hr}$$

$$v \approx 15 \text{ km/hr}$$

17. Consider two wires of same material having their ratio of radii to be 2 : 1. If these two wires are stretched by equal force, the ratio of stress produced in them is-

(A)  $\frac{1}{4}$  (B)  $\frac{1}{2}$  (C)  $\frac{3}{4}$  (D) 1

Ans. [A]

Sol. Stress =  $\frac{\text{Force}}{\text{Area}} \propto \frac{1}{\text{Area}}$

$$\text{Stress} \propto \frac{1}{r^2} \quad (\text{Area} = \pi r^2)$$

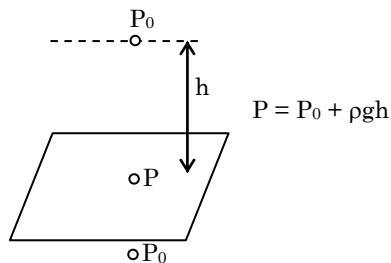
$$\text{ratio of stress} = \left(\frac{1}{2}\right)^2 = \frac{1}{4}$$

18. A submarine has a window of area  $30 \times 30 \text{ cm}^2$  on its ceiling and is at a depth of 100 m below sea level in a sea. If the pressure inside the submarine is maintained at the sea-level atmosphere pressure, then the force acting on the window is (consider density of sea water =  $1.03 \times 10^3 \text{ kg/m}^3$ , acceleration due to gravity =  $10 \text{ m/s}^2$ )

(A)  $0.93 \times 10^5 \text{ N}$  (B)  $0.93 \times 10^3 \text{ N}$  (C)  $1.86 \times 10^5 \text{ N}$  (D)  $1.86 \times 10^3 \text{ N}$

Ans. [A]

Sol.





$P$  = Pressure on upper surface of window

$$= P_0 + \rho gh$$

$P_{in}$  = Pressure inside the submarine

$$= P_0$$

Net force =  $(P_0 + \rho gh)A - P_0A$

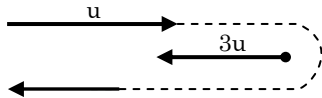
$$= \rho gh A$$

$$= 1.03 \times 10^3 \times 10 \times 100 \times 900 \times 10^{-4}$$

$$= 9.27 \times 10^4 \text{ Newton}$$

$$= 0.93 \times 10^5 \text{ Newton}$$

19. A spacecraft which is moving with a speed  $u$  relative to the earth in the  $x$ -direction, enters the gravitational field of a much more massive planet which is moving with a speed  $3u$  in the negative  $x$ -direction. The spacecraft exits following the trajectory as shown below.



The speed of the spacecraft with respect to the earth a long time after it has escaped the planet's gravity is given by

(A)  $u$

(B)  $4u$

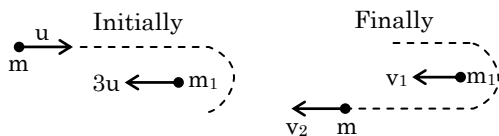
(C)  $2u$

(D)  $7u$

Ans.

[D]

Sol.



from momentum conservation

$$-mu + m_1 3u = m_1 v_1 + mv_2 \quad \dots(1)$$

from energy conservation

$$\frac{1}{2} mu^2 + \frac{1}{2} m_1 9u^2 = \frac{1}{2} m_1 v_1^2 + \frac{1}{2} mv_2^2$$

$$\frac{1}{2} mu^2 + \frac{1}{2} m_1 (3u - v_1)(3u + v_1) = \frac{1}{2} mv_2^2$$

from equation  $\dots(1)$

$$\Rightarrow m_1(3u - v_1) = m(v_2 + u)$$

$$\frac{1}{2} mu^2 + \frac{1}{2} m(v_2 + u)(3u + v_1) = \frac{1}{2} mv_2^2$$

as  $m_1 \gg m$ , we can assume  $v_1 \approx 3u$

$$u^2 + (v_2 + u)(6u) = v_2^2 \Rightarrow v_2 = 7u$$

20. The earth's magnetic field was flipped by  $180^\circ$  a million years ago. This flip was relatively rapid and took  $10^5$  years. Then the average change in orientation per year during the flip was closest to,

(A) 1 second

(B) 5 seconds

(C) 10 seconds

(D) 30 seconds

Ans.

[B]

Sol.

$$1^\circ = 3600 \text{ arc sec}$$

average change in orientation per year

$$= \frac{180^\circ}{10^5} \text{ degree/year}$$

$$= \frac{180 \times 3600}{10^5} \text{ sec/year}$$

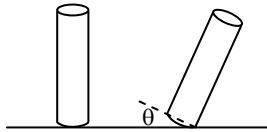
$$= 1.8 \times 0.36$$

$$= 6.48 \text{ sec/year}$$

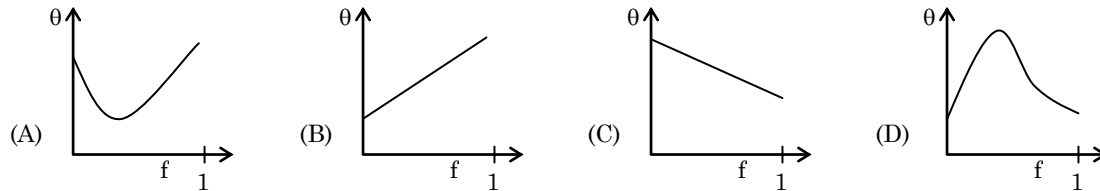
Closest option (B)



24. Figure below shows a shampoo bottle in a perfect cylindrical shape



In a simple experiment, the stability of the bottle filled with different amount of shampoo volume is observed. The bottle is tilted from one side and then released. Let the angle  $\theta$  depicts the critical angular displacement resulting in the bottle losing its stability and tipping over. Choose the graph correctly depicting the fraction  $f$  of shampoo filled ( $f = 1$  corresponds to completely filled) vs the tipping angle  $\theta$ .



Ans. [D]

Sol.

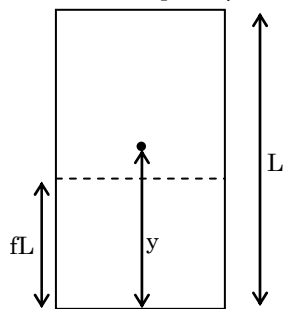
Mass of bottle =  $m_0$

Length of bottle =  $L$

base Area =  $A = \pi r^2$

density of shampoo =  $\rho$

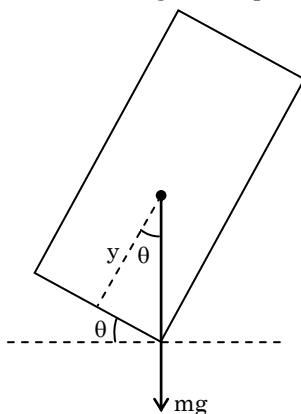
mass of shampoo =  $\rho fAL$



Center of mass of system

$$y = \frac{m_0 \frac{L}{2} + (\rho fAL) \left( \frac{fL}{2} \right)}{m_0 + \rho fAL}$$

for critical angular displacement,  $mg$  will pass through tilted side.



From the diagram  $\tan \theta = \frac{r}{y}$

$$\tan\theta = \frac{r(m_0 + \rho ALf)}{\frac{L}{2}(m_0 + \rho ALf^2)}$$

at  $f = 0$  and  $f = 1$ , tipping angle ' $\theta$ ' will be same.  
for very small values of  $f$ , we can neglect  $f^2$  terms

$$\Rightarrow \tan\theta = \frac{r}{\frac{L}{2}} \frac{(m_0 + \rho ALf)}{m_0}$$

$$\theta = \tan^{-1} \left( \frac{r}{\frac{L}{2}} \frac{(m_0 + \rho ALf)}{m_0} \right)$$

So if  $f$  increases  $\theta$  will increase.

25. At a height of 10 km above the surface of earth, the value of acceleration due to gravity is the same as that of a particular depth below the surface of earth. Assuming uniform mass density of the earth, the depth is,  
(A) 1 km (B) 5 km (C) 10 km (D) 20 km

Ans. [D]

Sol. At a height 'h'

$$g_h = g_0 \left( 1 - \frac{2h}{R} \right)$$

when  $h$  (10 km)  $<$   $R$  (6400 km)

at a depth 'd'

$$g_d = g_0 \left( 1 - \frac{d}{R} \right)$$

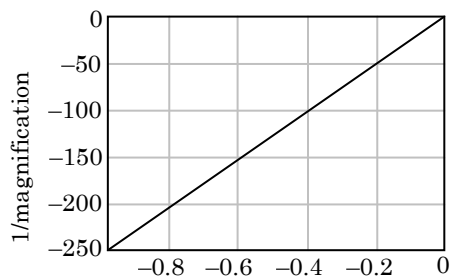
$$\text{here } g_0 = \frac{GM}{R^2}$$

Now  $g_h = g_d$

$$\Rightarrow d = 2h$$

$$d = 20 \text{ km}$$

26. The following graph depicts the inverse of magnification versus the distance between the object and lens data for a setup. The focal length of the lens used in the setup is



Distance between the object and the lens (m)

- (A) 250 m (B) 0.004 m (C) 125 m (D) 0.002 m

Ans. [B]

Sol. Magnification (m) =  $+\frac{v}{u}$

From lens formula,

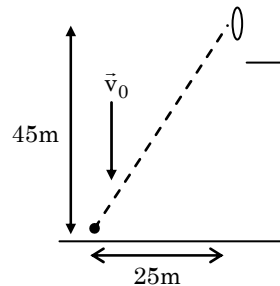
$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f} \Rightarrow \frac{u}{v} - 1 = \frac{u}{f} \Rightarrow \frac{u}{v} = \frac{u}{f} + 1$$

$\therefore$  graph between  $\left(\frac{u}{v}\right)$  [inverse of magnification] and  $u$  will be straight line with slope  $\frac{1}{f}$

From graph, slope = 250

$$\therefore f = \frac{1}{250} \text{ m} = 0.004 \text{ m}$$

27. In a circus, a performer throws an apple towards a hoop held at 45 m height by another performer standing on a high platform (see figure below). The thrower aims for the hoop and throws the apple with a speed of 24 m/s. At the exact moment that the thrower released the apple, the other performer drops the hoop. The hoop falls straight down. At what height above the ground does the apple go through the hoop?



- (A) 21 m                      (B) 22 m                      (C) 23 m                      (D) 24 m

Ans. [B]

Sol. Velocity of projection = 24 m/s

Distance between point of projection and hoop =  $\sqrt{25^2 + 45^2}$

$\therefore$  Time taken by ball to reach the hoop

$$= \frac{\sqrt{25^2 + 45^2}}{24}$$

(Note : We are analysing the motion wrt hoop)

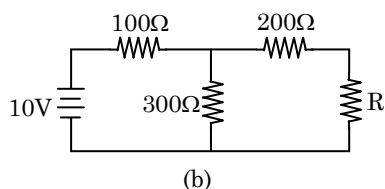
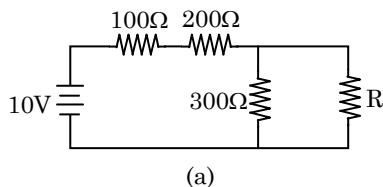
$\therefore$  Distance by which hoop will fall

$$= \frac{1}{2}at^2 = \frac{1}{2} \times 10 \times \frac{(25^2 + 45^2)}{24^2}$$

$\therefore$  Height above the ground where apple go through the hoop is given by

$$45 - \left[ \frac{1}{2} \times 10 \times \frac{(25^2 + 45^2)}{24^2} \right] = 22 \text{ m}$$

28. A student was trying to construct the circuit shown in the figure below marked (a), but ended up constructing the circuit marked (b) Realizing her mistake, she corrected the circuit, but to her surprise, the output voltage (across R) did not change.



The value of resistance R is-

- (A) 100  $\Omega$                       (B) 150  $\Omega$                       (C) 200  $\Omega$                       (D) 300  $\Omega$

Ans. [A]

Sol. For circuit (a),

$$i_R = \left( \frac{10}{\frac{300R}{300+R} + 300} \right) \times \frac{300}{300+R}$$

↑

Current through cell

[Note : 300 Ω and R are in parallel which is in series with 100 & 200 Ω]

$$\therefore V_{R_a} = \frac{10 \times 300 R}{300R + 300^2 + 300R}$$

[ $V_{R_a}$  is potential difference across resistance R]

For circuit (b),

$$i_R = \left( \frac{10}{\frac{(200+R)(300)}{200+R+300} + 100} \right) \times \frac{300}{300+200+R}$$

↑

Current through cell

[Note : R and 200 Ω are in series which is in parallel with 300 Ω and again the combination is in series with 100 Ω]

$$\therefore V_{R_b} = \frac{100 \times 300 R}{300 \times 200 + 300R + 100 \times 500 + 100R}$$

[ $V_{R_b}$  is potential difference across resistance R]

According to given situation

$$V_{R_a} = V_{R_b}$$

$$\therefore 300R + 9 \times 10^4 + 300R = 6 \times 10^4 + 400R + 5 \times 10^4$$

$$\Rightarrow 200R = 2 \times 10^4 \Rightarrow R = 100 \Omega$$

29. The ratio of gravitational force and electrostatic repulsive force between two electrons is approximately (gravitational constant =  $6.7 \times 10^{-11} \text{ Nm}^2/\text{kg}^2$ , mass of an electron =  $9.1 \times 10^{-31} \text{ kg}$ , charge on an electron =  $1.6 \times 10^{-19} \text{ C}$ )
- (A)  $24 \times 10^{-24}$                       (B)  $24 \times 10^{-36}$                       (C)  $24 \times 10^{-44}$                       (D)  $24 \times 10^{-54}$

Ans. [C]

Sol. Gravitational force  $F_G = \frac{Gm_1m_2}{r^2}$

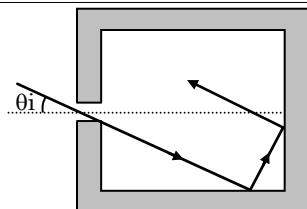
$$\text{Electrostatic force } F_e = \frac{1}{4\pi\epsilon_0} \frac{q_1q_2}{r^2}$$

$$\therefore \frac{F_G}{F_e} = \frac{Gm_1m_2 \cdot 4\pi\epsilon_0}{q_1q_2}$$

$$= \frac{6.7 \times 10^{-11} \times 9.1 \times 10^{-31} \times 9.1 \times 10^{-31}}{1.6 \times 10^{-19} \times 1.6 \times 10^{-19} \times 9 \times 10^9}$$

$$= 24 \times 10^{-44}$$

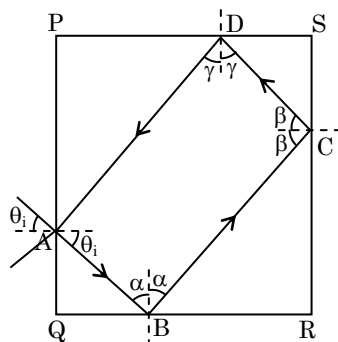
30. A monochromatic beam of light enters a square enclosure with mirrored interior surfaces at an angle of incidence  $\theta_i$  ( $\neq 0$ ) (see the figure below). For some value (s) of  $\theta_i$ , the beam is reflected by every mirrored wall (other than the one with opening) exactly once and exists the enclosure through the same hole. Which of the following statements about this beam is correct ?



- (A) The beam will not come out the enclosure for any value of  $\theta_i$   
 (B) The beam will not come out for more than two values of  $\theta_i$   
 (C) The beam will not come out only at  $\theta_i = 45^\circ$   
 (D) The beam will come out for exactly two values of  $\theta_i$

Ans. [C]

Sol.



$$\text{From geometry, } \alpha + \theta_i = \frac{\pi}{2} \Rightarrow \alpha = \frac{\pi}{2} - \theta_i$$

$$\beta + \alpha = \frac{\pi}{2} \Rightarrow \beta = \theta_i$$

$$\beta + \gamma = \frac{\pi}{2} \Rightarrow \gamma = \frac{\pi}{2} - \theta_i$$

Also  $AD \parallel BC$  and  $AB \parallel CD$

$\therefore$  ABCD is a parallelogram and  $AB = CD$

Also,  $\triangle ABQ \cong \triangle CDS$

$\therefore$  From trigonometry

$$\frac{AQ}{QB} = \frac{CR}{BR}$$

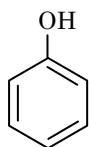
Let length of each side of square by  $\ell$ ,  $AQ = x$ , and  $QB = y$

$$\therefore \frac{x}{y} = \frac{\ell - x}{\ell - y} \Rightarrow x = y$$

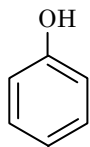
$$\therefore \theta_i = \frac{\pi}{4}$$

## CHEMISTRY

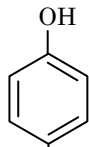
31. The acidity of



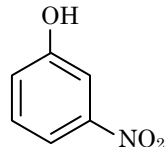
I



II



III



IV

Follows the order

(A) I > II > III > IV

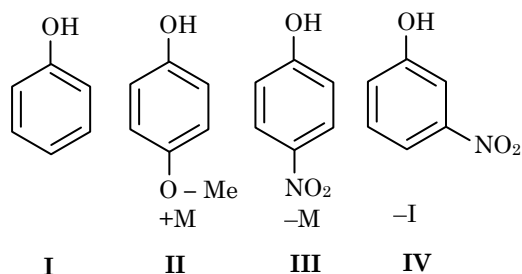
(B) IV > III > II > I

(C) III > IV > I > II

(D) III > II > IV > I

Ans. [C]

Sol.

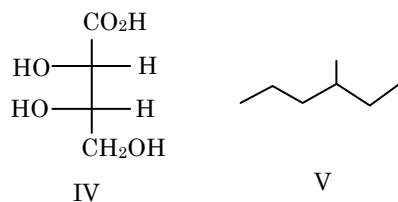
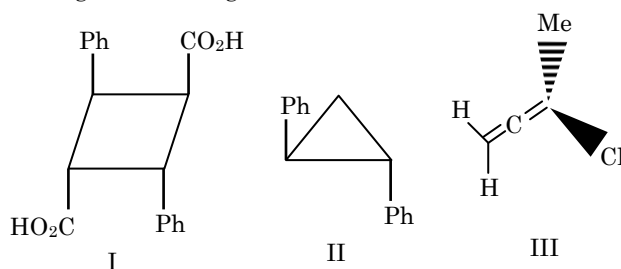


for acidic strength :

$-M > -HC > -I > - > +I > +HC > +M$

So III > IV > I > II

32. Among the following



the compounds which can exhibit optical activity are-

(A) only II, IV and V

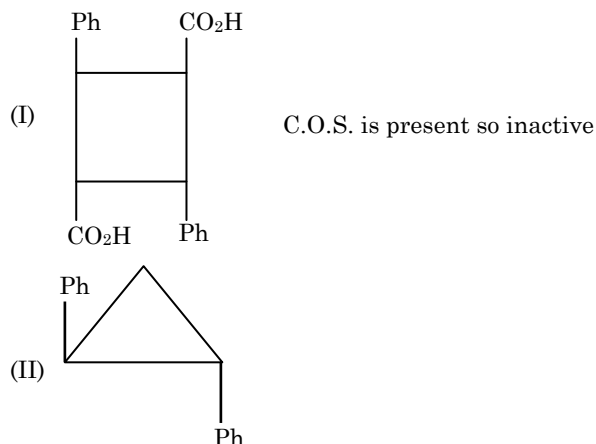
(B) only IV and V

(C) only I, II and V

(D) only I, II and IV

Ans. [A]

Sol.





C.O.S. → X  
 P.O.S. → X  
 A.A.O.S. → X  
 so active

(III) P.O.S. is present so inactive

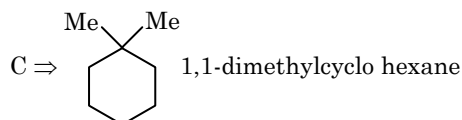
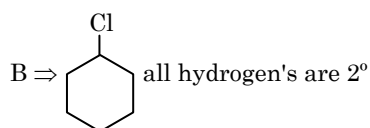
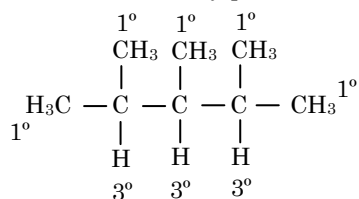
(IV) C.O.S and P.O.S. both are not present so active

(V) C.O.S and P.O.S. both are not present so active

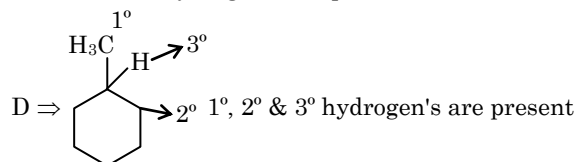
33. A molecule which has  $1^\circ$ ,  $2^\circ$  and  $3^\circ$  carbon atoms is-  
 (A) 2,3,4-trimethylpentane (B) chlorocyclohexane  
 (C) 2,2-dimethylcyclohexane (D) methylcyclohexane

Ans. [D]

Sol. A ⇒ 2,3,4-trimethylpentane only  $1^\circ$  and  $3^\circ$  hydrogen's are present



$1^\circ$  and  $2^\circ$  hydrogen's are present



34. The organic compound which can be purified by steam distillation is-  
 (A) acetone (B) aniline (C) glucose (D) ethanol

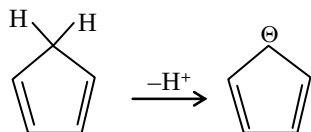
Ans. [B]

Sol. Aniline is purified by distillation method

35. Among the following, the most acidic compound is



Ans. [B]



Sol. Non aromatic                      Aromatic

36. A closed 10 L vessel contains 1 L water gas (1 : 1 CO : H<sub>2</sub>) and 9 L air (20% O<sub>2</sub> by volume) at STP. The contents of the vessel are ignited. The number of moles of CO<sub>2</sub> in the vessel is closest to-
- (A) 0.22                                      (B) 0.022                                      (C) 0.90                                      (D) 3.60

**Ans.** [B]

**Sol.** Water gas (CO : H<sub>2</sub> is 1 : 1) = 1 litre

Air = 9 litre

1 litre water gas at STP  $\Rightarrow \frac{1}{22.4}$  moles of gas at STP

No. of moles of CO =  $\frac{1}{2} \times \frac{1}{22.4}$  moles.

= No. of moles of CO<sub>2</sub> produced after ignition

= 0.022

37. A certain metal has a work function of  $\Phi = 2$  eV. It is irradiated first with 1 W of 400 nm light and later with 1 W of 800 nm light. Among the following, the correct statement is : [Given Planck constant (h) =  $6.626 \times 10^{-34}$  m<sup>2</sup> kgs<sup>-1</sup>; Speed of light (c) =  $3 \times 10^8$  ms<sup>-1</sup>]
- (A) Both colors of light give rise to same number of photoelectrons  
 (B) 400 nm light gives rise to less energetic photoelectrons than 800 nm light  
 (C) 400 nm light leads to more photoelectrons  
 (D) 800 nm light leads to more photoelectrons

**Ans.** [C]

**Sol.** Work function of metal ( $\phi$ ) = 2eV

Energy of photon ( $\lambda = 400$  nm) =  $\frac{hc}{\lambda} = 3.105$  eV

Energy of photon ( $\lambda = 800$  nm) =  $\frac{hc}{\lambda} = 1.5525$  eV

Hence, photon with  $\lambda = 400$  nm will emit photoelectrons while photon with  $\lambda = 800$  nm will not emit photoelectrons.

38. Among the following, the correct statement about the chemical equilibrium is
- (A) Equilibrium constant is independent of temperature  
 (B) Equilibrium constant is independent of temperature  
 (C) At equilibrium, the forward and the backward reactions stop so that the concentrations of reactants and products are constant  
 (D) Equilibrium constant is independent of whether you start the reaction with reactants or products

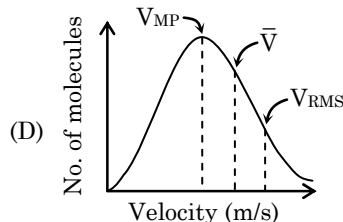
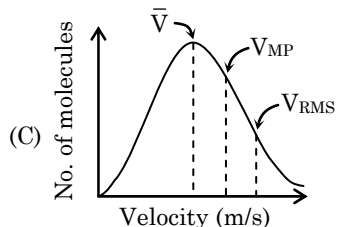
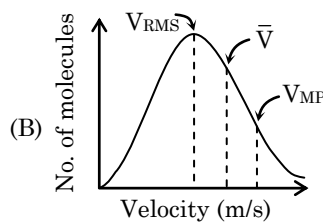
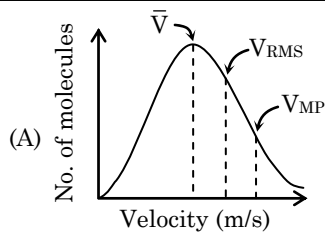
**Ans.** [D]

**Sol.** - Equilibrium constant is dependent on temperature

- Equilibrium constant do not tell us about the rate of reaction

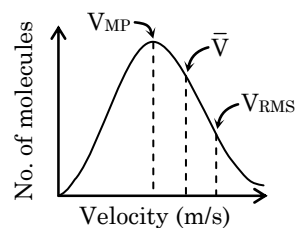
- At equilibrium, the forward and backward reactions do not stop but they have same rate

39. Among the following, the plot that shows the correct marking of most probable velocity ( $V_{MP}$ ), average velocity ( $\bar{V}$ ), and root mean square velocity ( $V_{RMS}$ ) is



Ans. [D]

Sol.



40. The correct set of quantum numbers for the unpaired electron of Cu atom is

(A)  $n = 3, \ell = 2, m = -2, s = +1/2$

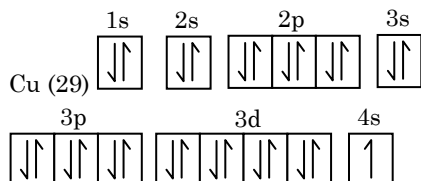
(B)  $n = 3, \ell = 2, m = +2, s = -1/2$

(C)  $n = 4, \ell = 0, m = 0, s = +1/2$

(D)  $n = 4, \ell = 1, m = +1, s = +1/2$

Ans. [C]

Sol. Cu [Ar]  $3d^{10} 4s^1$



The set of quantum numbers for the unpaired  $e^-$  of Cu atom is.

$$n = 4, \ell = 0, m = 0, s = +\frac{1}{2}$$

41. Among the following, the most polar molecule is

(A)  $AlCl_3$

(B)  $CCl_4$

(C)  $SeCl_6$

(D)  $AsCl_3$

Ans. [D]

Sol.

$AlCl_3$	non-polar
$CCl_4$	non-polar
$SeCl_6$	non-polar
$AsCl_3$	polar

42. The covalent characters of  $CaCl_2$ ,  $BaCl_2$ ,  $SrCl_2$  and  $MgCl_2$  follow the order

(A)  $CaCl_2 < BaCl_2 < SrCl_2 < MgCl_2$

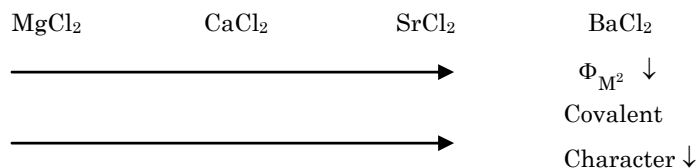
(B)  $BaCl_2 < SrCl_2 < CaCl_2 < MgCl_2$

(C)  $CaCl_2 < BaCl_2 < MgCl_2 < SrCl_2$

(D)  $SrCl_2 < MgCl_2 < CaCl_2 < BaCl_2$

Ans. [B]

Sol.

BaCl<sub>2</sub> < SrCl<sub>2</sub> < CaCl<sub>2</sub> < MgCl<sub>2</sub> (Covalent character)

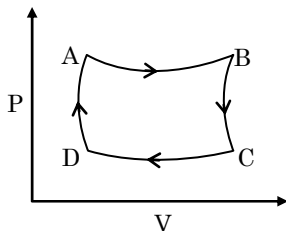
43. Among the following, the correct statement is-

- (A) 100 has four significant figures  
 (B)  $1.00 \times 10^2$  has four significant figures  
 (C) 2.005 has four significant figures  
 (D) 0.0025 has four significant figures

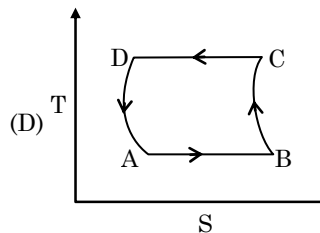
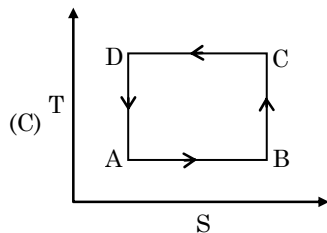
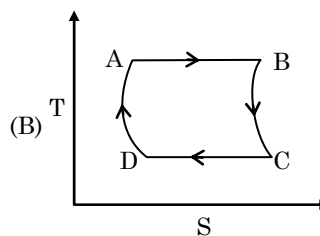
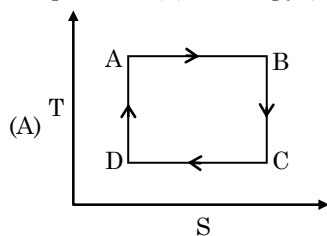
Ans. [C]

Sol.

44. A thermodynamic cycle in the pressure (P) – volume (V) plane is given below :



AB and CD are isothermal processes while BC and DA are adiabatic processes. The same cycle in the temperature (T) – entropy (S) plane is-



Ans. [A]

Sol. AB is isothermal reversible expansion process i.e.  $\Delta T = 0$  and S increases as there is increase in volume.  
 BC is adiabatic reversible expansion process ( $q_{rev} = 0$ ) i.e. temperature decreases and  $\Delta S = 0$ .  
 CD is isothermal reversible compression process i.e.  $\Delta T = 0$  and S decreases as there is decrease in volume.  
 DA is adiabatic reversible compression process ( $q_{rev} = 0$ ) i.e. temperature decreases and  $\Delta S = 0$ .

45. The first ionization potential (IP) of the elements Na, Mg, Si, P, Cl and Ar are 5.14, 7.65, 8.15, 10.49, 12.97 and 15.76 eV, respectively. The IP (in eV) of K is closest to-

(A) 13.3                      (B) 18.2                      (C) 4.3                      (D) 6.4

Ans. [C]

Sol. The first ionisation potential of K is less than Na.  
 $\therefore$  The first ionization potential of K is closest to 4.3.

## BIOLOGY

46. Which ONE of the following chemicals serves as a substrate for carbonic anhydrase ?  
 (A) O<sub>2</sub> (B) CO<sub>2</sub> (C) NO<sub>2</sub> (D) CO

Ans. [B]

Sol.

47. Which ONE of the following is NOT a function of the small intestine ?

- (A) Absorption of end products of digestion  
 (B) Digestion of proteins  
 (C) Digestion of lipids  
 (D) Acidification of ingested food

Ans. [D]

Sol.

48. Insulin stimulates the conversion of glucose to  
 (A) fructose (B) glycogen (C) sucrose (D) starch

Ans. [B]

Sol.

49. Which ONE of the following statements about ecosystem energetics is INCORRECT ?  
 (A) The metabolic requirements of poikilotherms are higher than that of homeotherms  
 (B) Autotrophs form the base of the food chain in natural ecosystems  
 (C) In terrestrial ecosystems, most of the primary production is consumed by detritivores and not herbivores  
 (D) Approximately 10% energy of one trophic level is transferred to the next level

Ans. [A]

Sol.

50. Proton motive force is created by pumping protons across the  
 (A) *trans*-Golgi network (B) endoplasmic reticulum  
 (C) mitochondrial inner membrane (D) early endosomal membrane

Ans. [C]

Sol.

51. Which ONE of the following Mendelian diseases is an example of X-linked recessive disorder ?  
 (A) Haemophilia (B) Phenylketonuria  
 (C) Sickle cell anaemia (D) Beta-thalassaemia

Ans. [A]

Sol.

52. Which ONE of the following pairs gives rise to fruit and seed, respectively, in a typical angiosperm plant ?  
 (A) Ovule and ovary (B) Ovary and pollen  
 (C) Pollen and anther (D) Ovary and ovule

Ans. [D]

Sol.

53. The concept of vaccinating arose from Edward Jenner's observation that
- (A) injecting inactivated anthrax spores in sheeps protected them from anthrax  
 (B) injecting humans with tuberculosis-infected lung extracts protected them from tuberculosis  
 (C) milk-maids previously infected with cowpox did not contract small pox  
 (D) injecting inactivated rabies virus in humans protected them from rabies

Ans. [C]

Sol.

54. A plant with genotype *AABBCC* is crossed with another plant with *aabbcc* genotype. How many different genotypes of pollens is possible in an F<sub>1</sub> plant if these three loci follow independent assortment ?
- (A) 8 (B) 4 (C) 2 (D) 1

Ans. [A]

Sol.

55. Which ONE of the following sequences of events CORRECTLY represents mitosis ?
- (A) Metaphase, telophase, prophase, anaphase  
 (B) Anaphase, prophase, metaphase, telophase  
 (C) Prophase, anaphase, metaphase, telophase  
 (D) Prophase, metaphase, anaphase, telophase

Ans. [D]

Sol.

56. The amount of air that is left behind in lungs after expiratory reserve volume has been exhaled is
- (A) inspiratory reserve volume (B) tidal volume  
 (C) residual volume (D) vital capacity

Ans. [C]

Sol.

57. Match the species in Column-I with their respective feature of body organisation in Column-II.

Column-I	Column-II
P. Mollusca	i. Pseudocoelom
Q. Annelida	ii. Radula
R. Nematoda	iii. Radial symmetry
S. Echinodermata	iv. Segmentation

Choose the CORRECT combination.

- (A) P-ii, Q-i, R-iv, S-iii (B) P-ii, Q-iv, R-i, S-iii  
 (C) P-iii, Q-iv, R-i, S-ii (D) P-iv, Q-iii, R-ii, S-i

Ans. [B]

Sol.

58. Who among the following scientists proposed the theory of natural selection independently of Charles Darwin?
- (A) Alfred Russel Wallace (B) Carl Linnaeus  
 (C) Georges Cuvier (D) Jean-Baptiste Lamarck

Ans. [A]

Sol.

59. The maximum concentration of harmful chemical is expected to be found in organisms  
 (A) at the bottom of a food chain (B) at the middle of a food chain  
 (C) at the top of a food chain (D) at any level in a food chain

Ans. [C]

Sol.

60. The genome of SARS-CoV2 is composed of  
 (A) double stranded DNA (B) double stranded RNA  
 (C) single stranded DNA (D) single stranded RNA

Ans. [D]

Sol.

## Part - II

### Two - Mark Questions

## MATHEMATICS

61. Let A denote the set of all 4-digit natural numbers with no digit being 0. Let  $B \subset A$  consist of all numbers x such that no permutation of the digits of x gives a number that is divisible by 4. Then the probability of drawing a number from B with all even digits is  
 (A)  $\frac{625}{1641}$  (B)  $\frac{16}{641}$  (C)  $\frac{16}{1641}$  (D)  $\frac{1000}{1641}$

Ans. [C]

Sol. All even digit numbers in  $B = \{2, 4, 6, 8\}$

fav : forming a 4 digit nos with all digit is even and not divisible by 4

2222, 6666, 2266,

$$\frac{4!}{2!2!} = 6 \text{ cases}$$

(2226), (2666)

$$\frac{4!}{3!} = 4 \text{ cases} \quad \frac{4!}{3!} = 4 \text{ cases}$$

Total = 1 + 1 + 6 + 4 + 4 = 16 cases

Total : forming a 4-digit no. from  $\{1, 2, \dots, 9\}$  but not divisible by 4

C-1 all 4 digit are odd =  $5^4 = 625$

C-2 all 4 digit are even = 16

C-3 one even and 3 odd

you can not take 2 or 6 as one of even digit

(12, 16, 32, 36, ..... ) all are divisible by 4

But you can take 4 or 8 as one of even digit .....

$${}^2C_1 \times {}^4C_1 \times 5^3 = 1000$$

Take one even digit out of 4 & 8

select one place for 4 & 8 out of 4-places

$$\text{Total} = 1000 + 625 + 16 = 1641$$

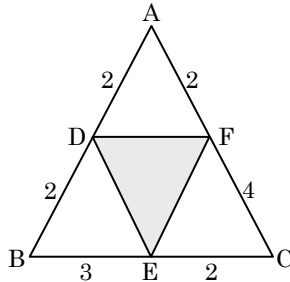
$$P = \frac{\text{fav.}}{\text{total}} = \frac{16}{1641}$$

62. Let ABC be a triangle such that  $AB = 4$ ,  $BC = 5$  and  $CA = 6$ . Choose points D, E, F on AB, BC, CA respectively, such that  $AD = 2$ ,  $BE = 3$ ,  $CF = 4$ . Then  $\frac{\text{area } \triangle DEF}{\text{area } \triangle ABC}$  is-

- (A)  $\frac{1}{4}$                       (B)  $\frac{3}{15}$                       (C)  $\frac{4}{15}$                       (D)  $\frac{7}{30}$

Ans. [C]

Sol.



$$\frac{\text{area } (\triangle DEF)}{\text{area } (\triangle ABC)} =$$

Let  $\text{area } (\triangle ABC) = \Delta$

$$\frac{\text{area } (\triangle BED)}{\text{area } (\triangle ABC)} = \frac{\frac{1}{2} \cdot 6 \sin B}{\frac{1}{2} \cdot 20 \sin B} = \frac{3}{10}$$

$$\frac{\text{area } (\triangle ADF)}{\text{area } (\triangle ABC)} = \frac{\frac{1}{2} \cdot 4 \sin A}{\frac{1}{2} \cdot 24 \sin A} = \frac{1}{6}$$

$$\frac{\text{area } (\triangle CEF)}{\text{area } (\triangle ABC)} = \frac{\frac{1}{2} \cdot 8 \sin C}{\frac{1}{2} \cdot 30 \sin C} = \frac{4}{15}$$

$$\text{area } (\triangle BED) = \frac{3}{10} \Delta$$

$$\text{area } (\triangle ADF) = \frac{\Delta}{6}$$

$$\text{area } (\triangle CEF) = \frac{4\Delta}{15}$$

$$\text{area } (\triangle DEF) = \Delta - \left( \frac{3}{10} + \frac{1}{6} + \frac{4}{15} \right) \Delta$$

$$= \Delta - \left( \frac{27 + 15 + 24}{90} \right) \Delta$$

$$= \Delta - \left( \frac{66}{90} \right) \Delta = \frac{24\Delta}{90}$$

$$\frac{\text{area } (\triangle DEF)}{\text{area } (\triangle ABC)} = \frac{24}{90} = \frac{4}{15}$$

63. The number of ordered pairs  $(x, y)$  of integers satisfying  $x^3 + y^3 = 65$  is-

- (A) 0                      (B) 2                      (C) 4                      (D) 6

Ans. [B]

Sol.  $x^3 + y^3 = 65$



$$\begin{aligned}
 (x+y)(x^2+y^2-xy) &= 65 \times 1 \\
 &= 13 \times 5 \\
 &= 5 \times 13 \\
 &= 1 \times 65
 \end{aligned}$$

clearly  $x^2 + y^2 - xy > 0$

C-1 :  $x + y = 5$  and  $x^2 + y^2 - xy = 13$

$$x^2 + (5-x)^2 - x(5-x) = 13$$

$$3x^2 - 15x + 12 = 0$$

$$x^2 - 5x + 4 = 0 \Rightarrow x = 1, 4$$

$$(x, y) = (1, 4) \text{ and } (4, 1)$$

C-2 :  $x + y = 13$  and  $x^2 + y^2 - xy = 5$

$$x^2 + (13-x)^2 - x(13-x) = 5$$

$$3x^2 - 39x + 164 = 0, x \notin \mathbb{I} \text{ (Not possible)}$$

C-3 :  $x + y = 1$  and  $x^2 + y^2 - xy = 65$

$$x^2 + (1-x)^2 - x(1-x) = 65$$

$$3x^2 - 3x - 64 = 0, x \notin \mathbb{I} \text{ (Not possible)}$$

C-4 :  $x + y = 65$  and  $x^2 + y^2 - xy = 65$

No solution

so two ordered pair satisfy the relation

64. A bottle in the shape of a right-circular cone with height  $h$  contains some water. When its base is placed on a flat surface, the height of the vertex from the water level is  $a$  units. When it is kept upside down, the height of the base from the water level is  $\frac{a}{4}$  units. Then the ratio  $\frac{h}{a}$  is

(A)  $\frac{1+\sqrt{85}}{4}$

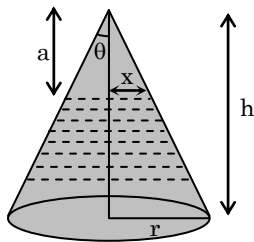
(B)  $\frac{1+\sqrt{85}}{8}$

(C)  $\frac{1+\sqrt{65}}{4}$

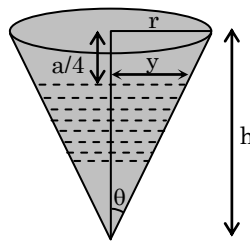
(D)  $\frac{1+\sqrt{65}}{8}$

Ans. [B]

Sol.



$$\frac{r}{h} = \frac{x}{a} \Rightarrow x = \frac{ra}{h}$$



$$\frac{r}{h} = \frac{y}{h - \frac{a}{4}}$$

$$y = \frac{r}{h} \left( h - \frac{a}{4} \right)$$

Equating volume of water in both cases

$$\frac{1}{3}(\pi r^2 h - \pi x^2 a) = \frac{1}{3}\pi y^2 \left( h - \frac{a}{4} \right)$$

$$\Rightarrow r^2 h - r^2 h - \frac{r^2 a^2}{h} \cdot a = \frac{r^2}{h^2} \left( h - \frac{a}{4} \right)^2 \left( h - \frac{a}{4} \right)$$

$$\Rightarrow \frac{h^2}{a^2} - \frac{h}{4a} - \frac{21}{16} = 0$$

$$\frac{h}{a} = \frac{\frac{1}{4} \pm \sqrt{\frac{1}{16} + \frac{21}{4}}}{2}$$

$$\frac{h}{a} = \frac{1 + \sqrt{85}}{8}$$

65. Consider the following two statements :

I. if  $n$  is a composite number, then  $n$  divides  $(n - 1)!$

II. There are infinitely many natural numbers  $n$  such that  $n^3 + 2n^2 + n$  divides  $n!$

Then

(A) I and II are true

(B) I and II are false

(C) I is true and II is false

(D) I is false and II is true

Ans. [D]

Sol. If  $n$  is a composite number (Take  $n = 4$ )

For  $n = 4$ ,  $n$  does not divide  $(n - 1)!$

Hence Ist statement is false

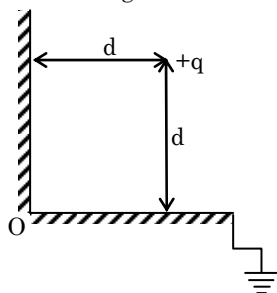
(II)  $n^3 + 2n^2 + n = n(n + 1)^2$

$n(n + 1)^2$  divides  $n!$

If  $n$  is such that  $(n + 1)$  is a prime number (Take  $n = 6$ ) so  $n(n + 1)^2$  does not divide  $n!$  but there are infinite values of  $n$  ( $n = 104, 109, 114, \dots$ ) for which  $n(n + 1)^2$  divide  $n!$  but it is not true for every natural numbers.

## PHYSICS

66. A charge  $+q$  is situated at a distance ' $d$ ' away from both the sides of a grounded conducting 'L' shaped sheet as shown in the figure.



The force acting on the charge  $+q$  is-

(A) towards O, magnitude  $\frac{q^2}{32\pi\epsilon_0 d^2} (2\sqrt{2} + 1)$

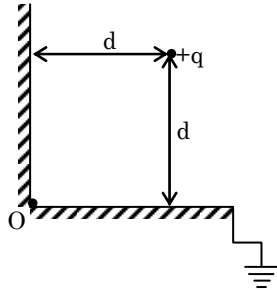
(B) away from O, magnitude  $\frac{q^2}{32\pi\epsilon_0 d^2} (2\sqrt{2} + 1)$

(C) towards O, magnitude  $\frac{q^2}{32\pi\epsilon_0 d^2} (2\sqrt{2} - 1)$

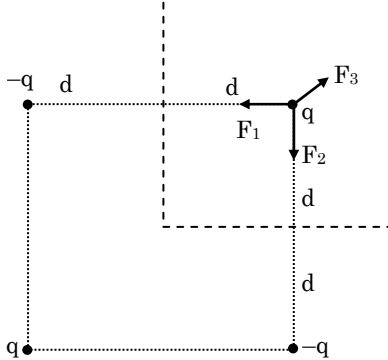
(D) away from O, magnitude  $\frac{q^2}{32\pi\epsilon_0 d^2} (2\sqrt{2} - 1)$

Ans. [C]

Sol.



By method of image, the given arrangement is equivalent to

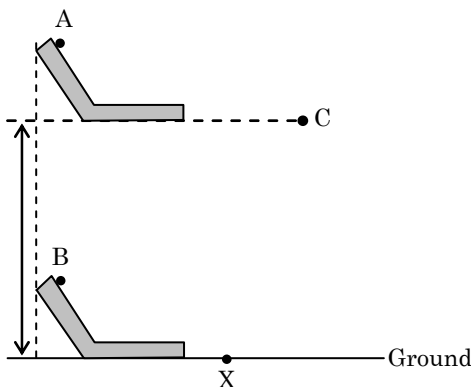


$$F_1 = \frac{1}{4\pi\epsilon_0} \frac{q^2}{(2d)^2}, F_2 = \frac{1}{4\pi\epsilon_0} \frac{q^2}{(2d)^2}$$

$$F_3 = \frac{1}{4\pi\epsilon_0} \frac{q^2}{(2\sqrt{2}d)^2}$$

$$\begin{aligned} \therefore F_{\text{net}} &= \sqrt{2} \frac{q^2}{16\pi\epsilon_0 d^2} - \frac{q^2}{32\pi\epsilon_0 d^2} \\ &= \frac{q^2}{32\pi\epsilon_0 d^2} (2\sqrt{2} - 1) \text{ [towards O]} \end{aligned}$$

67. Three balls, A, B and C, are released and all reach the point X (shown in the figure). Balls A and B are released from two identical structures, one kept on the ground and the other at height h, from the ground as shown in the figure. They take time  $t_A$  and  $t_B$  respectively to reach X (time starts after they leave the end of the horizontal portion of the structure). The ball C is released from a point at height h, vertically above X and reaches X in time  $t_C$ . Choose the correct statement.



- (A)  $t_C < t_A < t_B$       (B)  $t_C = t_A = t_B$       (C)  $t_C = t_A < t_B$       (D)  $t_C < t_A = t_B$

Ans. [B]

Sol. Work done by gravity on A and B is same.

$\therefore$  Horizontal velocity of A = horizontal velocity of B as they leave the horizontal portion of the structure.

$$\therefore t_A = t_B \dots(i)$$

Also vertical velocity of A and vertical velocity of C when released are both zero

$\therefore$  They both will cover same vertical distance in same time.

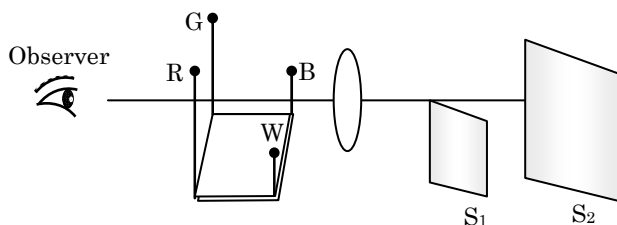
$$\therefore t_A = t_C \dots(ii)$$

From (i) and (ii)

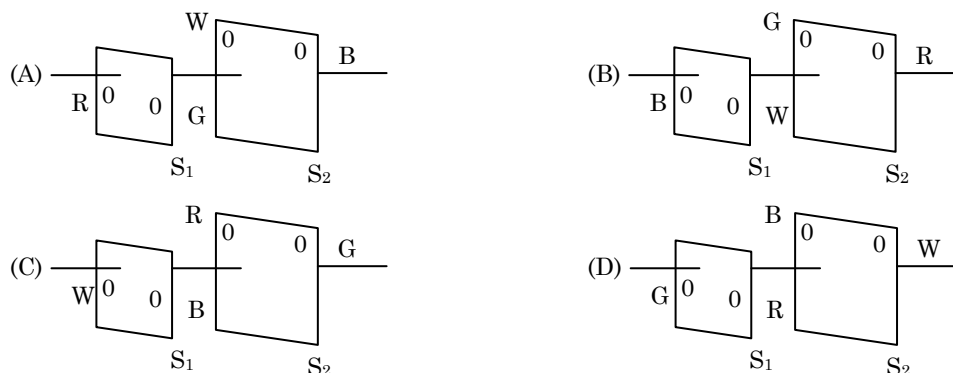
$$t_A = t_B = t_C$$

Note : Correction in optio. Option B should be  $t_A = t_B = t_C$

68. Four bulbs, red, green, white and blue (denoted by R, G, W and B respectively) are kept in front of a converging lens (as shown in the figure below). The observer sees that the green and blue bulbs are kept to the left of the principle axis while the red and white bulbs are kept to the right of the principle axis. He also see that the red and green bulbs are above the principle axis while the white and blue bulbs are below the principle axis. The screens  $S_1$  and  $S_2$  are set at appropriate positions for the focusing to view the images.



Choose the figure that correctly represents the images are seen by the observer.



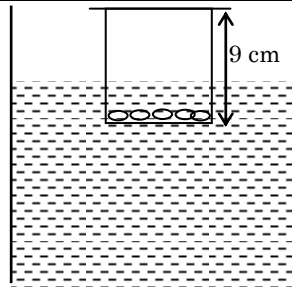
Ans. [A]

Sol. Since the images are being made on screen, hence real.

$\therefore$  Image will be inverted

Also since blue and white are nearer to lens, hence their real image will be far from lens as compared to red and green.

69. A wide bottom cylindrical massless plastic container of height 9 cm has 40 identical coins inside it and is floating on water with 3 cm inside the water. If we start putting more of such coins on its lid, it is observed that after N coins are put, it equilibrium changes from stable to unstable. Equilibrium in floating is stable if the geometric center of the submerged portion is above the center of mass of the object). The value of N is closest to



- (A) 6                                      (B) 10                                      (C) 16                                      (D) 24

Ans. [B]

Sol. Let mass of each coin be  $m$ .

$\therefore$  Location of center of mass after  $N$  coins are kept on lid from bottom of container is

$$\frac{40m \times 0 + Nm \times 9}{(40 + N)m} = \frac{9N}{40 + N} \text{ cm}$$

Also height of submerged portion after keeping  $N$  coins on lid will be,

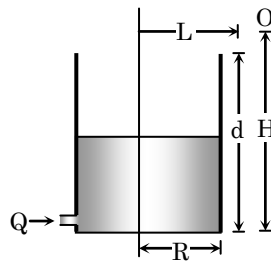
$$\frac{4(40 + N)}{40} \text{ cm}$$

$\therefore$  Equilibrium will just be stable if

$$\frac{3}{40} \frac{(40 + N)}{2} = \frac{9N}{(40 + N)}$$

$$\Rightarrow 3N^2 - 480N + 4800 = 0 \Rightarrow N = 10.72$$

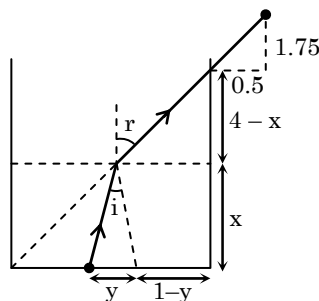
70. A small coin is fixed at the center of the base of an empty cylindrical steel container having radius  $R = 1\text{ m}$  and height  $d = 4\text{ m}$ . At time  $t = 0\text{ s}$ , the container starts getting filled with water at a flow rate of  $Q = 0.1\text{ m}^3/\text{s}$  without disturbing the coin. Find the approximate time when the coin will first be seen by the observer "O" from the height of  $H = 5.75\text{ m}$  above and  $L = 1.5\text{ m}$  radially away from the coin as shown in the figure. Refractive index of water is  $n = 1.33$ .



- (A) 0 s                                      (B) 32 s                                      (C) 63 s                                      (D) 150 s

Ans. [C]

Sol.

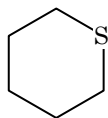


$$1.33 \sin i = \sin r = \frac{2}{\sqrt{53}} \dots(i)$$



Ans. [D]

Sol.  $0.233 \text{ gm BaSO}_4$  has 1 millimole  $\text{BaSO}_4$  and hence has 1 millimole S  
 $\therefore$  organic compound (X) also has 1 millimole S % of S in  $0.102 \text{ gm}$  of organic compound (X)  
 $= \frac{0.032}{0.102} \times 100 = 31.37 \%$



102 gm of this organic compound has 32 gm S

This has same % of S.

73. The specific heat of a certain substance is  $0.86 \text{ J g}^{-1} \text{ K}^{-1}$ . Assuming ideal solution behavior, the energy required (in J) to heat 10 g of 1 molal of its aqueous solution from 300 K to 310 K is closest to [Given : molar mass of the substance =  $58 \text{ g mol}^{-1}$ ; specific heat of water =  $4.2 \text{ J g}^{-1} \text{ K}^{-1}$ ]  
 (A) 401.7 (B) 424.7 (C) 420.0 (D) 86.0

Ans. [A]

Sol. Specific heat capacity of substance  
 $= 0.86 \text{ J g}^{-1} \text{ K}^{-1}$   
 1 molal aqueous solution  
 $\Rightarrow 1000 \text{ gm}$  water has 58 gm solute (total mass of solution = 1058 gm)  
 If we take 10 gm solution it would have

$$\text{water} = \frac{1000}{1058} \times 10 \text{ gm}$$

$$\text{substance} = \frac{58}{1058} \times 10 \text{ gm}$$

$$\text{Heat required} = \frac{1000}{1058} \times 10 \times 4.2 \times 10 \text{ (for water)} = 396.975$$

+

$$\frac{58}{1058} \times 10 \times 0.86 \times 10 \text{ (for substance)} = 4.715$$

$$= 396.975 + 4.715 = 401.69 \approx 401.7$$

74. Strength of a  $\text{H}_2\text{O}_2$  solution is labelled as 1.79 N. Its strength can also be expressed as closest to-  
 (A) 20 volume (B) 5 volume (C) 10 volume (D) 15 volume

Ans. KVPY Given [C]

Sol.  $2\text{H}_2\text{O}_{2(\text{aq.})} \rightarrow 2\text{H}_2\text{O}(\ell) + \text{O}_2(\text{g})$

It is a common trend that n-factor ( $\text{H}_2\text{O}_2$ ) is taken as '2'

By the definition of volume strength of  $\text{H}_2\text{O}_2$  if we consume 1  $\ell$  of 1 N  $\text{H}_2\text{O}_2$  in the above equation we are using 1 gram equivalent of  $\text{H}_2\text{O}_2 = 0.5$  moles of  $\text{H}_2\text{O}_2$  (by using n-factor = 2)

This will produce  $\frac{1}{4}$  moles of  $\text{O}_2$  gas at N.T.P.

$$\equiv \frac{1}{4} \times 22.4 = 5.6 \ell \text{ of } \text{O}_2 \text{ gas}$$

i.e. 1  $\ell$ , 1 N  $\text{H}_2\text{O}_2$  solution gives 5.6  $\ell$   $\text{O}_2$  at N.T.P.

Hence 1 N = 5.6 'vol.'  $\text{H}_2\text{O}_2$  solution

In the given question it is 1.76 N  $\text{H}_2\text{O}_2$  solution

Hence volume strength =  $5.6 \times 1.79 \approx 10$  volumes.

By Career Point Ans. (A)

$2\text{H}_2\text{O}_{2(\text{aq.})} \rightarrow 2\text{H}_2\text{O}(\ell) + \text{O}_2(\text{g})$

The above equation is a classic example of disproportionation of  $\text{H}_2\text{O}_2$ , hence we should take n-factor of  $\text{H}_2\text{O}_2 = 1$   
 $1 \ell, 1 \text{ N } \text{H}_2\text{O}_2$  solution  $\equiv 1$  gm equivalent of  $\text{H}_2\text{O}_2$

$$\equiv 1 \text{ mole of } \text{H}_2\text{O}_2$$

This will produce  $\frac{1}{2}$  moles of  $\text{O}_2$  gas at N.T.P.

$$= \frac{1}{2} \times 22.4 = 11.2 \ell \text{ of } \text{O}_2 \text{ gas}$$

i.e.,  $1 \ell, 1 \text{ N } \text{H}_2\text{O}_2$  solution gives  $11.2 \ell$  of  $\text{O}_2$  gas at N.T.P.

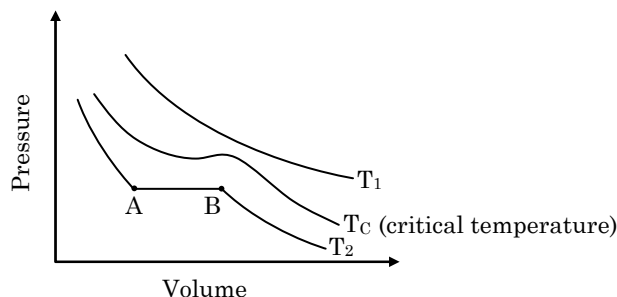
Hence  $1 \text{ N} \equiv 11.2$  'vol'  $\text{H}_2\text{O}_2$  solution

In the given question it is  $1.79 \text{ N } \text{H}_2\text{O}_2$  solution

Hence, volume strength =  $1.79 \times 11.2 \approx 20$  volumes

Suggested Answer is option (A)

75. The isotherms of a gas are shown below :



Among the following,

- (i) At  $T_1$ , the gas cannot be liquefied
- (ii) At point B, liquid starts to appear at  $T_2$
- (iii)  $T_c$  is the highest temperature at which the gas can be liquefied
- (iv) At point A, a small increase in pressure condenses the whole system to a liquid

The correct statements are :

- (A) only (i) and (ii)
- (B) only (i), (iii) and (iv)
- (C) only (ii), (iii) and (iv)
- (D) (i), (ii), (iii) and (iv)

Ans. [D]

Sol. Since  $T_1 > T_c$ , the gas cannot be liquefied at  $T_1$ .  $T_c$  is the highest temperature at which the gas can be liquefied.

At temperature  $T_2$ , liquid starts to appear at point B, however a small increase in pressure at point A condenses the whole system to liquid.

## BIOLOGY

76. Anthropocene refers to the geological age during which
- (A) the earliest hominids radiated from their ancestral forms
  - (B) human activity significantly influenced climate and environment
  - (C) arthropod radiation was highest
  - (D) arthropod radiation significantly influenced climate and environment

Ans. [B]

Sol.



77. Match the vitamins listed in Column-I with the diseases caused due to their deficiency in Column II.

Column-I	Column-II
P. Vitamin A	i. Pellegra
Q. Vitamin B <sub>2</sub>	ii. Rickets
R. Vitamin D	iii. Ariboflavinosis
S. Vitamin B <sub>12</sub>	iv. Pernicious anaemia

Choose the CORRECT combination

- (A) P-iv; Q-ii; R-iii; S-v  
 (B) P-i; Q-ii; R-iv; S-iii  
 (C) P-iv; Q-iii; R-ii; S-v  
 (D) P-iii; Q-iv; R-v; S-i

Ans. [C]

Sol.

78. An adult mammal with 50kg body weight has the following functional parameters of its lungs.

Inspiratory reserve volume = 40 ml/kg body weight

Expiratory reserve volume = 15 ml/kg body weight

Vital capacity = 60 ml/kg body weight

Breathing rate = 20/min

The volume (in litre) of air that its lungs displace in 24 hours is-

- (A) 72,000                      (B) 7,200                      (C) 3,600                      (D) 1,200

Ans. [B]

Sol.

79. In a breed of dog, long-haired phenotype is recessive to short-hair. In a litter, one pup is short-haired and its sibling is long-haired. Consider following possible phenotypes of the parents.

- i. Both parents are short-haired  
 ii. Both parents are long-haired  
 iii. One parent is short-haired, and one is long-haired

Choose the CORRECT combination of the possible parental phenotypes.

- (A) i only                      (B) ii only                      (C) iii only                      (D) i or iii

Ans. [D]

Sol.

80. In medical diagnostics for a disease, *sensitivity* (denoted by  $a$ ) of a test refers to the probability that a test result is positive for person with the disease, whereas *specificity* (denoted by  $b$ ) refers to the probability that person without the disease tests negative. A diagnostic test for COVID-19 has the values of  $a = 0.99$  and  $b = 0.99$ . If the prevalence of COVID-19 in a population is estimated to be 10%, what is the probability that a randomly chosen person tests positive for COVID-19 ?

- (A) 0.099                      (B) 0.10                      (C) 0.108                      (D) 0.11

Ans. [C]

Sol.